



Cedre Information Day

Modelling chemical dispersion

Paris, 27 March 2013

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1. Introduction

- *Context*
- *Phenomenological description*

2. Dispersion mechanisms in the aquatic environment

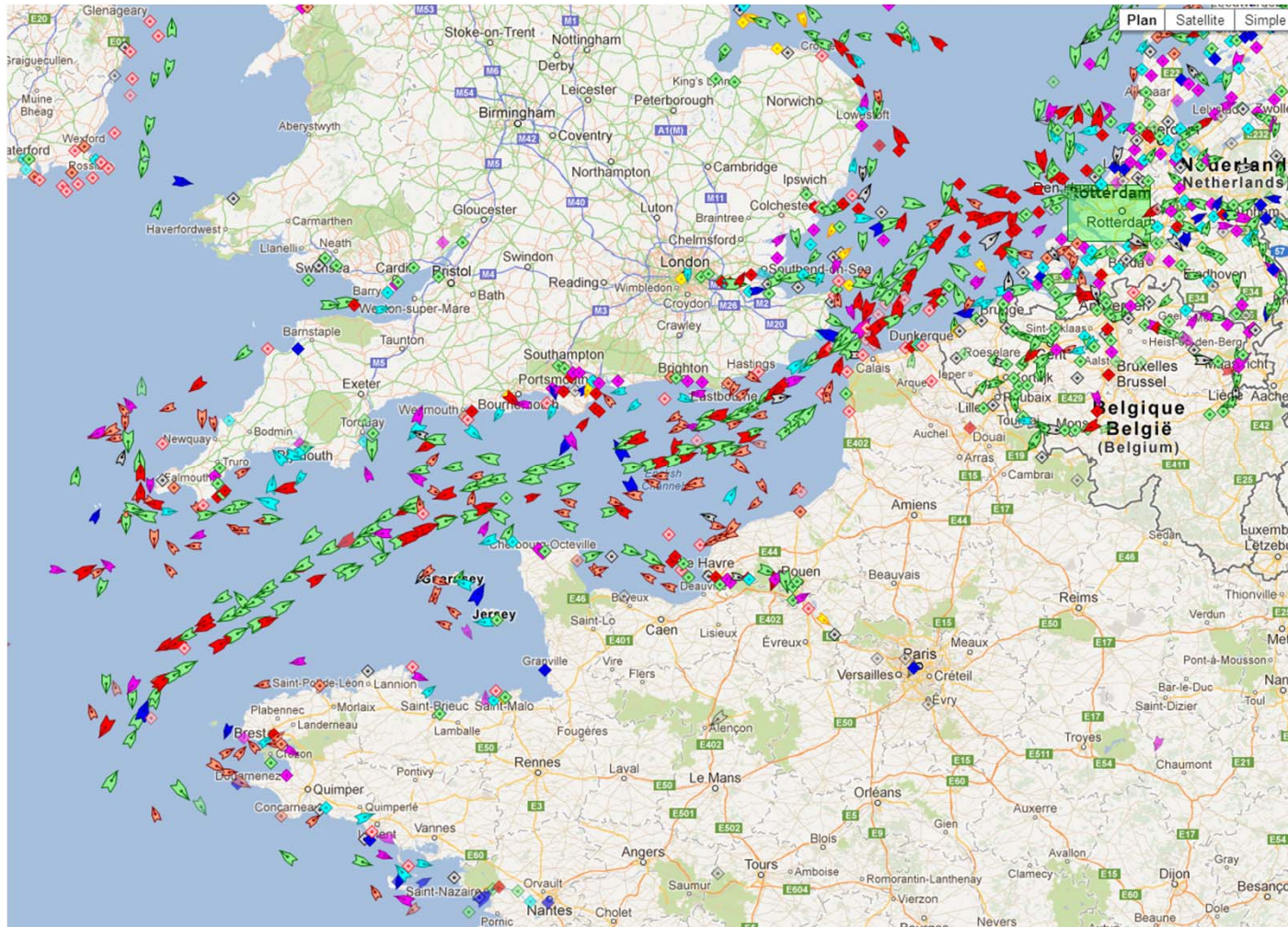
3. Modelling strategies

- *Surface release*
- *Subsurface release*

4. Example of modelling



Context



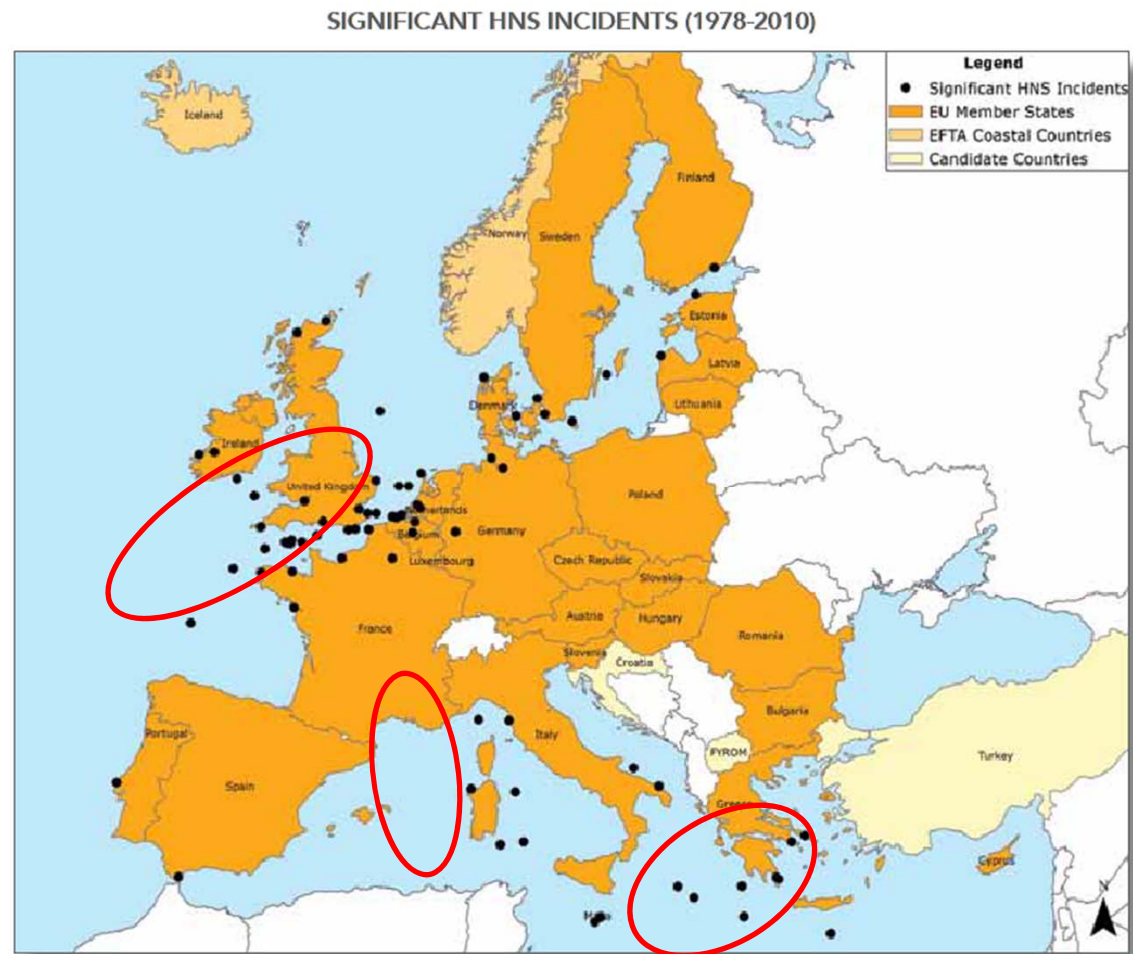
MarineTraffic.com



Context



Significant shipping incidents involving Hazardous and Noxious Substances (HNS)



4

Source: European Maritime Safety Agency (EMSA)

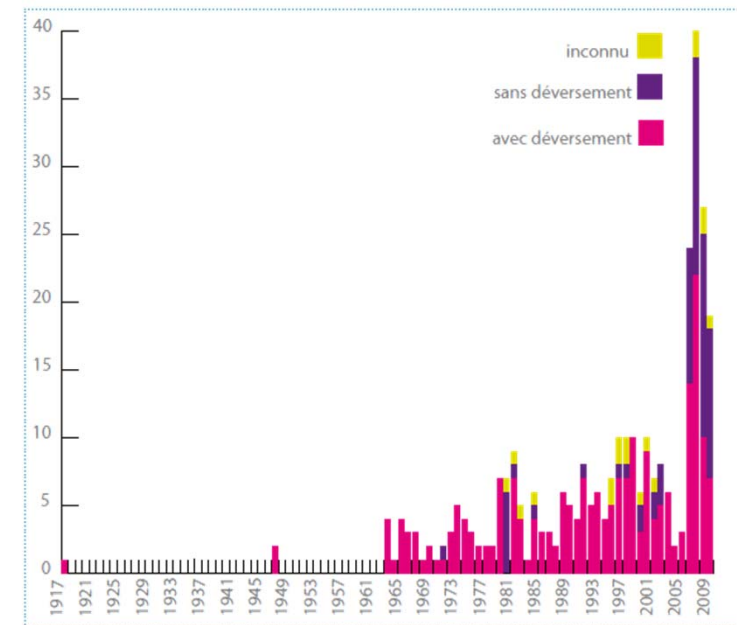


Context



Evolution of shipping incidents involving HNS

- ▶ Constant rise in chemical transport by sea
- ▶ 3.5-fold rise in chemical shipping in 20 years
- ▶ Wide variety of chemicals
- ▶ Different physical and chemical behaviour in the marine environment → SEBC code



Number of accidents involving HNS

Increase in the probability of incidents with severe consequences



Context



Wide variety of chemicals transported



Different response strategies



Pollutant characteristics

- ▶ Oils
- ▶ Chemicals
- ▶ Food products



Characteristics of polluted areas

- ▶ Sea, ocean, estuaries...
- ▶ Marine reserves, protected areas
- ▶ Aquaculture, fishing or tourism areas



Context

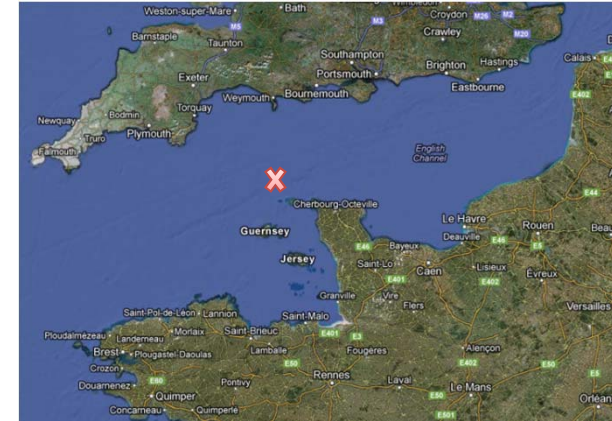


Example of an incident

levoli Sun (2000)

- ▶ Quantities transported
 - ▶ Styrene: 4000 T
 - ▶ Methyl Ethyl Ketone (MEK): 1000 T
 - ▶ Iso Propyl Alcohol (IPA): 1000 T

- ▶ Quantities released:
 - ▶ Styrene: 400 T
 - ▶ Methyl Ethyl Ketone: 100 T
 - ▶ Iso Propyl Alcohol: 1000 T



Leak in the bow thruster compartment



Context



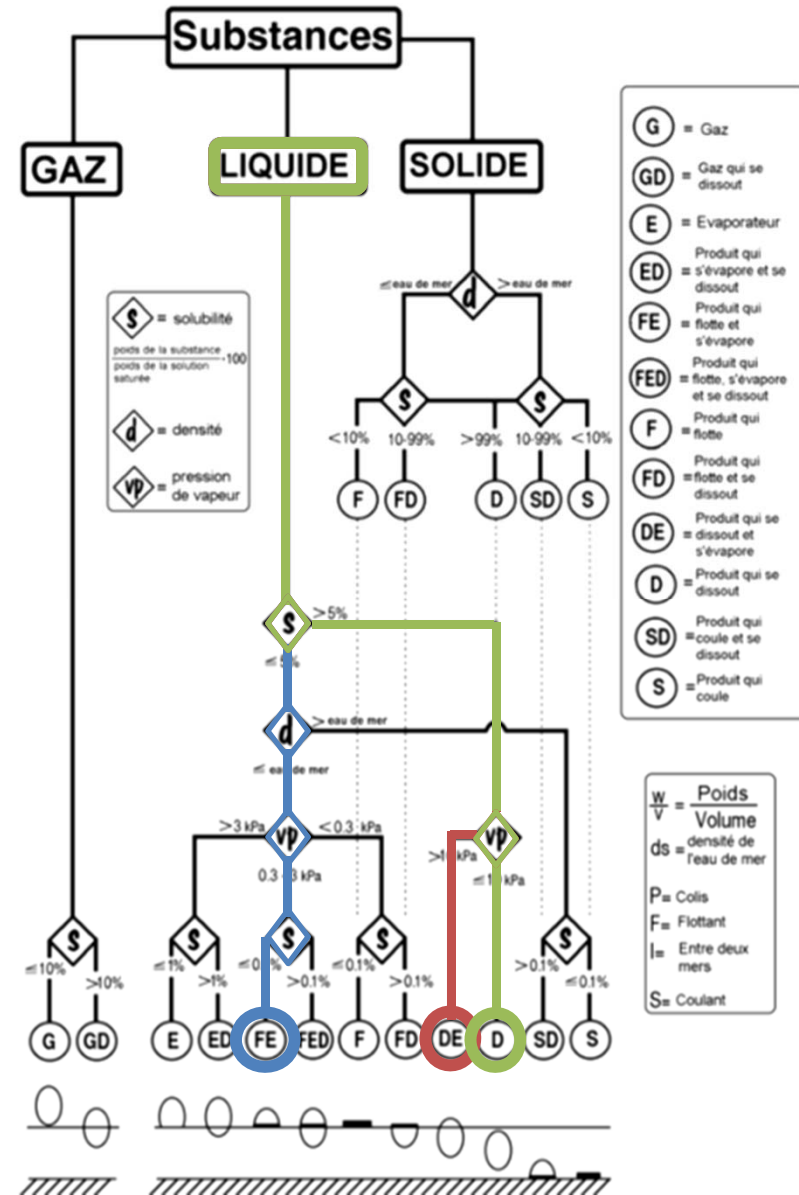
Example of an incident

levoli Sun (2000)

	Phase	S [%]	d [kg/m ³]	Vp [kPa]	
• Styrene	L	0.03	906	0.667	FE
• MEK	L	26.3	805	10.33	DE
• IPA	L	79	786	4.1	D



- ▶ Formation of a toxic cloud?
- ▶ Risk of fire or explosion?





Phenomenological description



Classification conditions

- ▶ Pure substance
- ▶ Small quantity
- ▶ 20°C
- ▶ Atmospheric pressure
- ▶ Fresh water

Real conditions

- ▶ Not necessarily pure
- ▶ Several tonnes
- ▶ 4°C
- ▶ 10 bars
- ▶ Seawater

Limitations

- ▶ Non-representative conditions
- ▶ Possible presence of different products
 - ▶ Alteration of dissolution
 - ▶ Chemical reaction
- ▶ No consideration of dynamics
 - ▶ Dissolution
 - ▶ Competition between phenomena



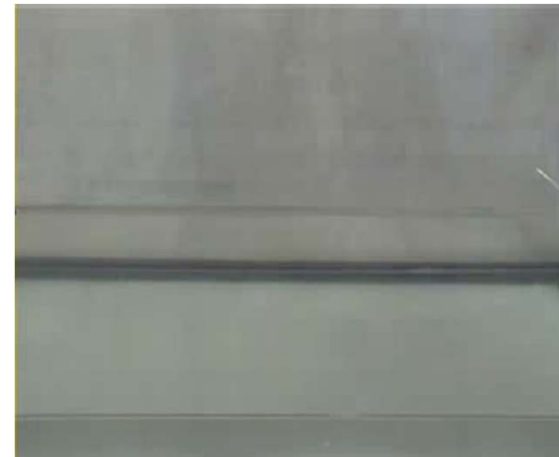
Phenomenological description



Example of behaviours



Sinker (e.g. Toluene Diisocyanate)



Evaporator (e.g. Styrene)



Floater (e.g. Dodecylbenzene)



Dissolver (e.g. butanol)



Dispersion mechanisms



Surface release

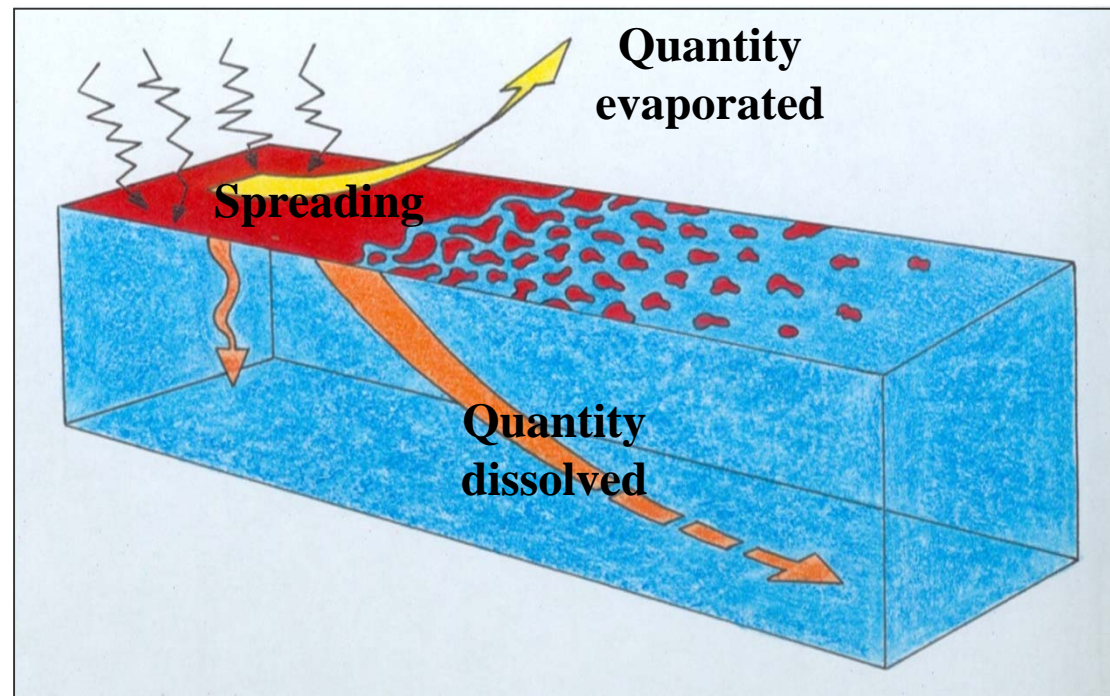
1. Determination of source term (quantity released or rate)-> Bernoulli
2. Material balance according to time

Surface area of the product?

Quantity evaporated?

Quantity dissolved?

How long?





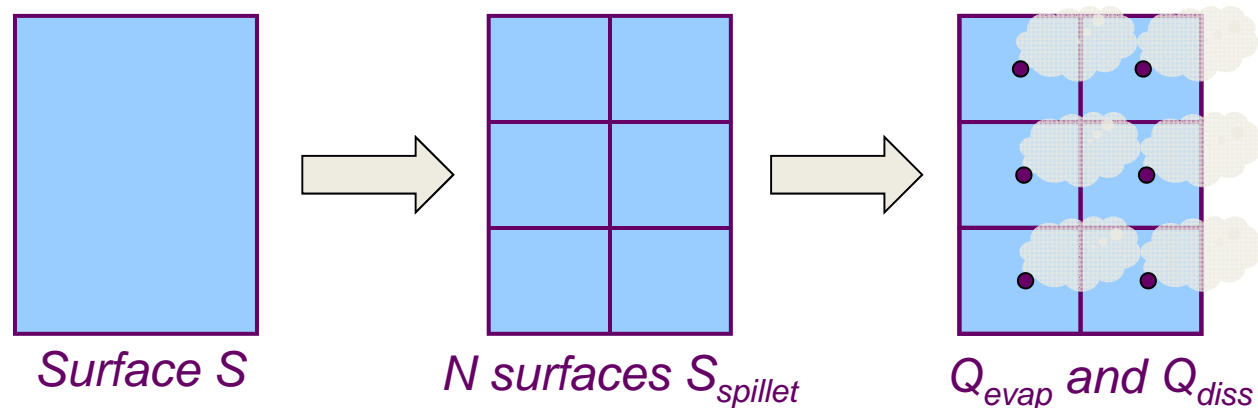
Modelling



Modelling

Division of slick into unitary elements (spillet) to represent the total surface area of the slick

1. Calculation of the evolution of each spillet over time (Runge-Kutta discretization scheme)
 - Spreading rate of each spillet (*Nihoul, 1983*)
 - Quantity evaporated (*Mackay and Matsugu, 1973*)
 - Quantity dissolved (*Hayduk and Laudie, 1974*)
2. Material balance for each spillet





Modelling



Modelling

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 - Spreading rate of each spillet (*Nihoul, 1983*)
 - Quantity evaporated (*Mackay and Matsugu, 1973*)
 - Quantity dissolved (*Hayduk and Laudie, 1974*)
2. Material balance for each spillet
3. Coupling with hydrodynamic predictions
 1. Movement of spilletts at each time step





Dispersion mechanisms



Subsurface release

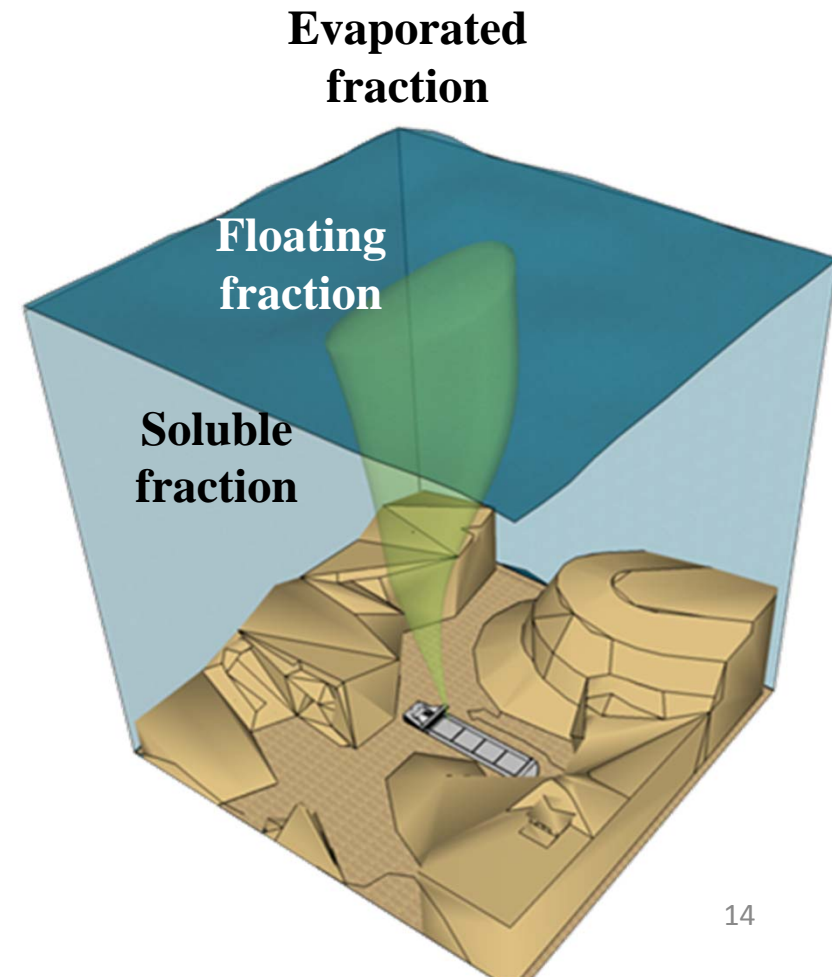
1. Determination of source (quantity released or rate)-> Bernoulli
2. Product dispersion mechanism
3. Upwelling hydrodynamics (speed)
4. Material transfer (dissolution)

Appearance of substance at surface

Volume at the surface?

When?

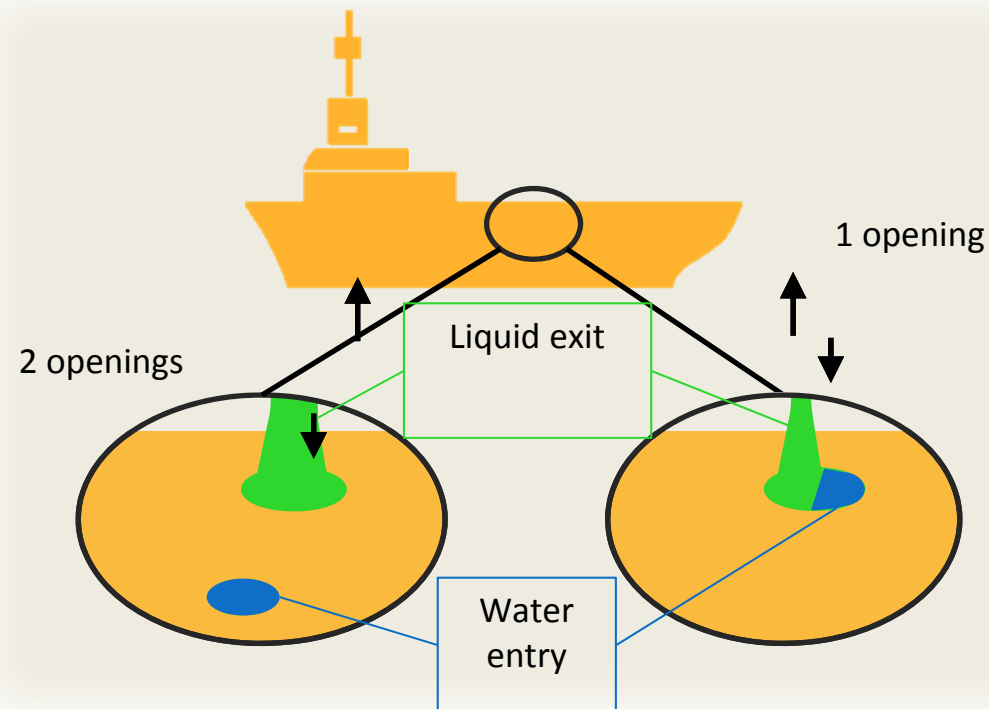
How long?





Modelling release rate at breach

1. Two hydrodynamic behaviours



2. Modelling by a local energy balance (Bernoulli)



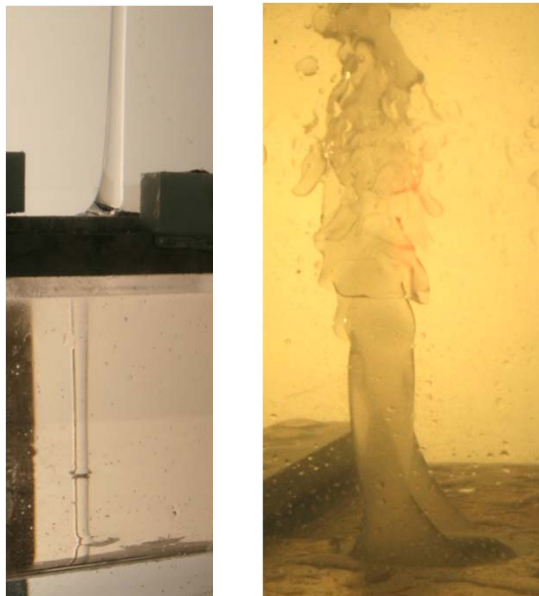
Modelling



Modelling release rate at breach

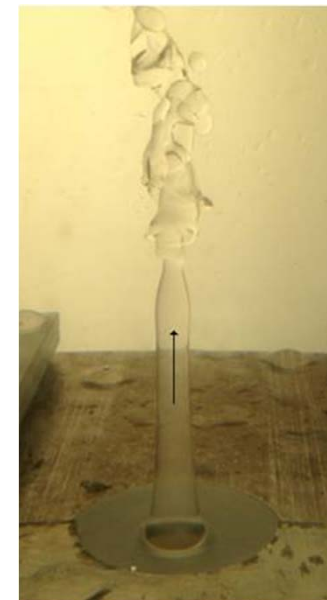
Constant rate Q ,
Linear section Σ

Diphasic release at the exit
opening



Linear rate Q ,
Constant section Σ

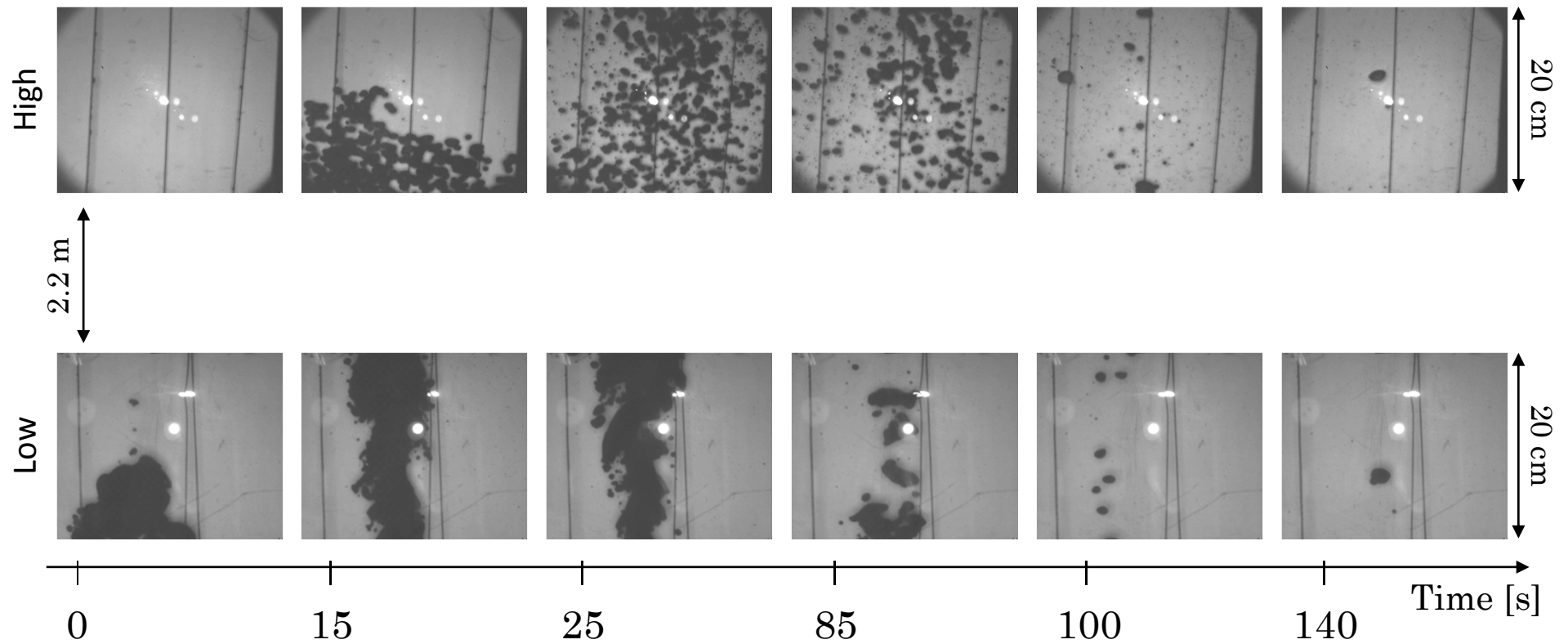
Monophasic release at the exit
opening





Modelling fragmentation of the leak

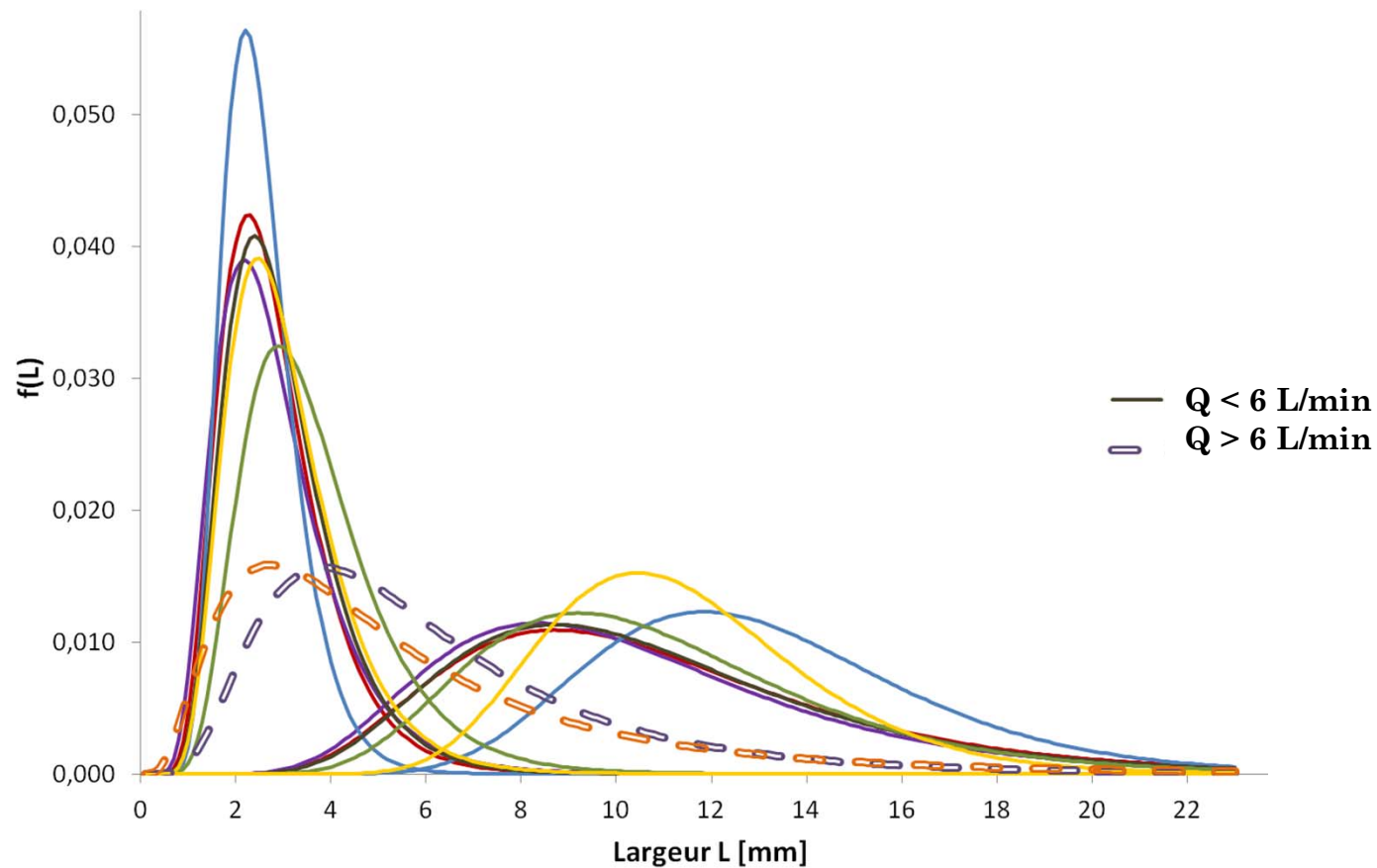
Assessment of particle size distribution





Modelling fragmentation of the leak

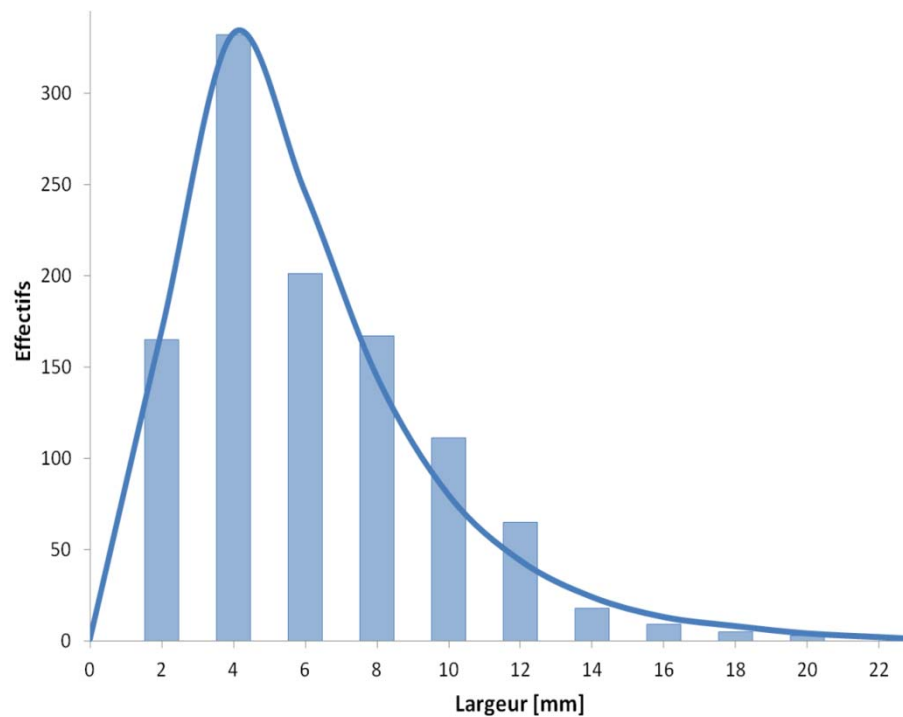
- ▶ 2 types of particle size distributions
- ▶ Maximum droplet diameter 22 mm



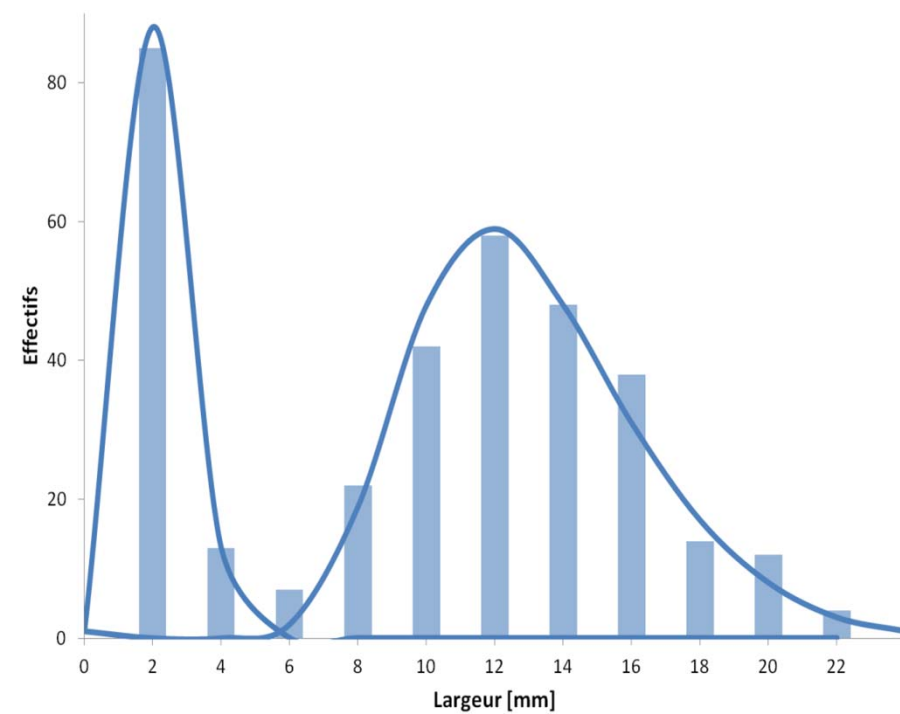


Modelling fragmentation of the leak

- Representation of particle size distributions by log-normal
 1. Monophasic release: monomodal distribution
 2. Diphasic release: bimodal distribution



Monophasic release

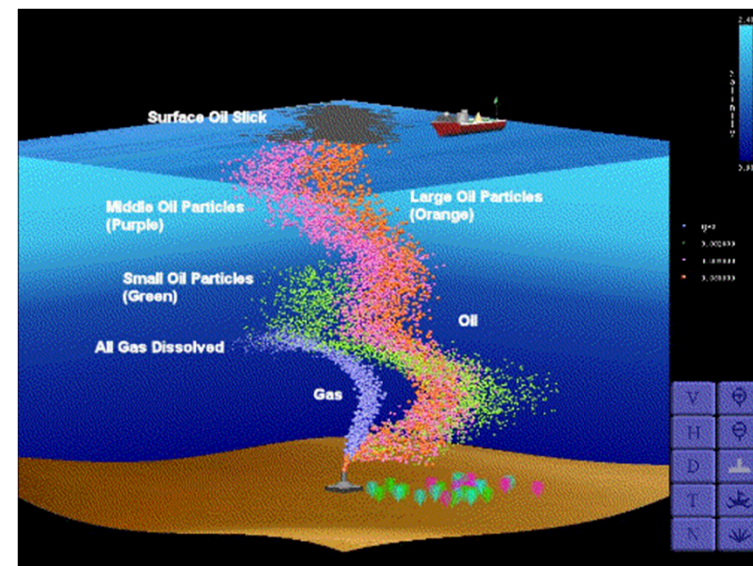
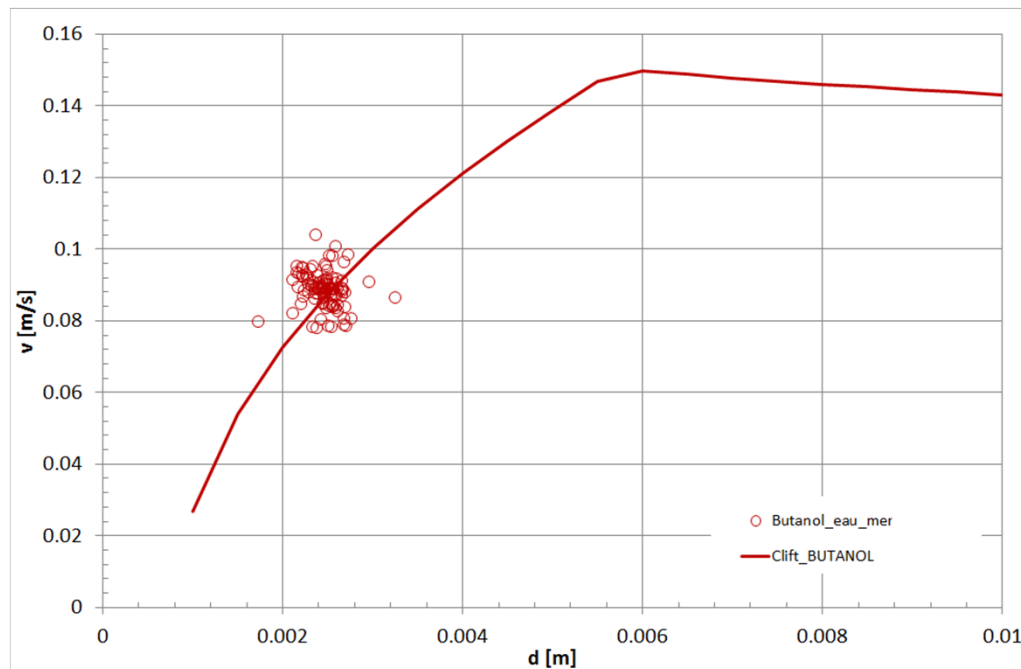


Diphasic release



Modelling droplet hydrodynamics

1. Modelling droplet upwelling speed with the Clift correlation (1978)
2. Coupling with hydrodynamic predictions to take into account the movement of the bubble plume



CDOG model, Clarkson University

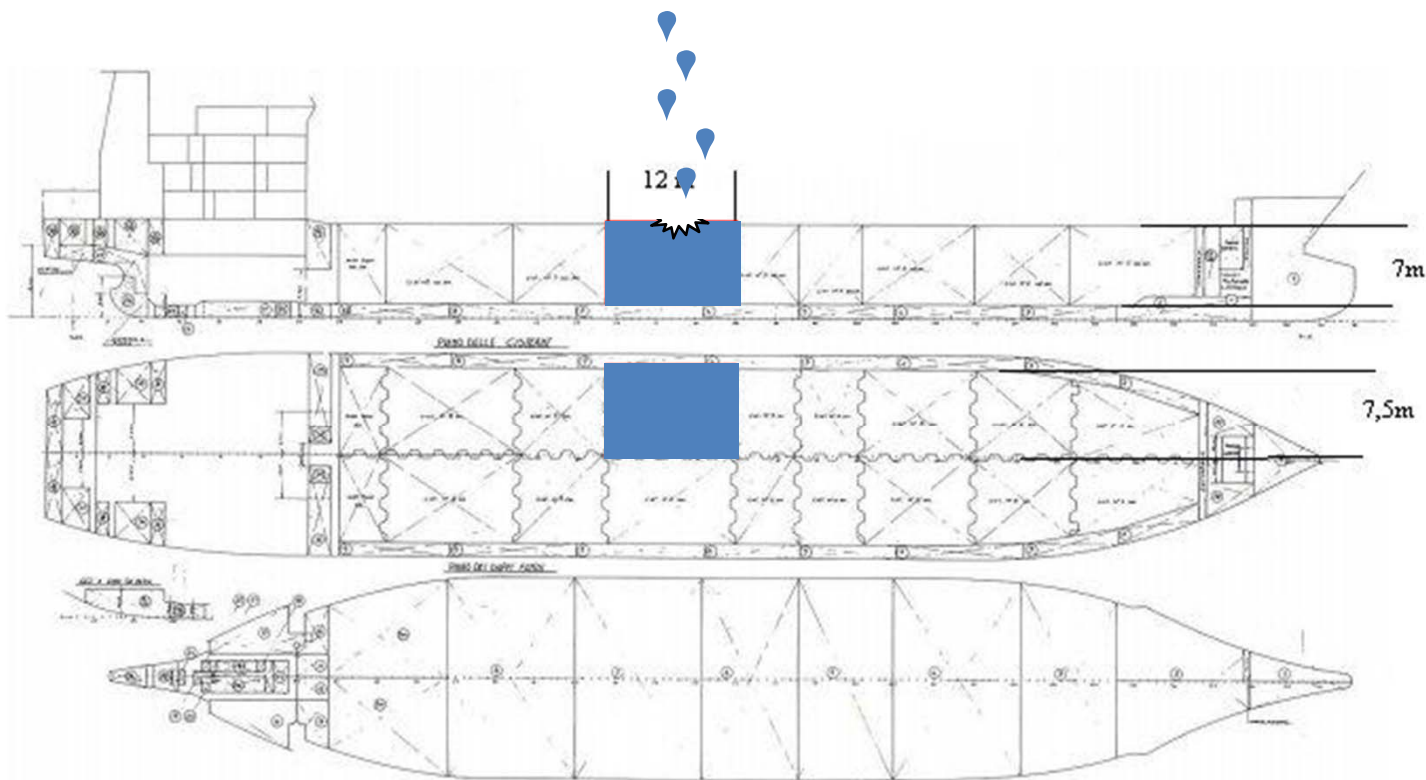


Example



Modelling a leak of butanol from a chemical tanker

1. Leak from a 600 m³ tank
2. Breach punctured along 10 cm in diameter



Source: BEAMer

CEPPOL: Centre d'Expertises Pratiques de Lutte Antipollution

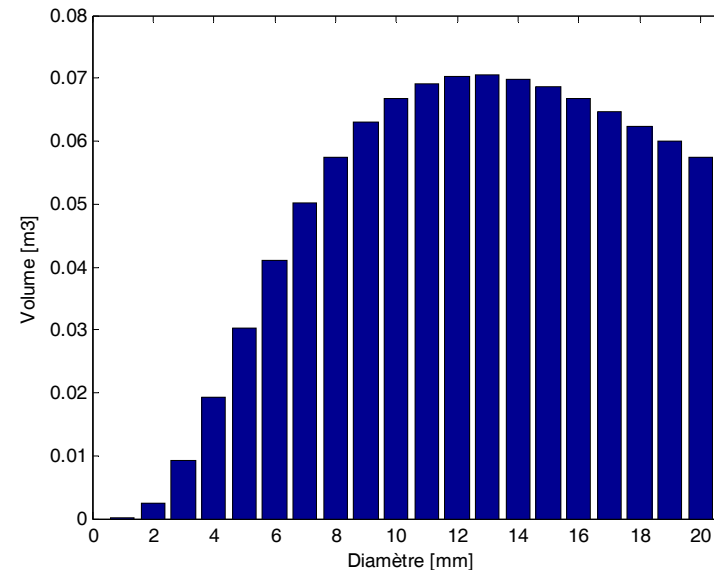
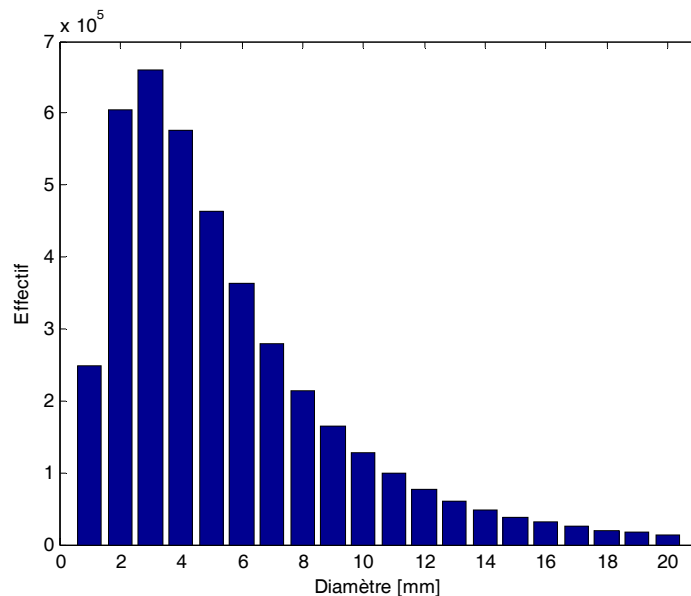


Example



Modelling a leak of butanol from a chemical tanker

1. Modelling the leak rate: tank emptied in 1h44
2. Dispersion in the water column
 1. Number of droplets per diameter group for 1m³ of substance released
 2. Distribution of volume per group for 1m³ of substance released





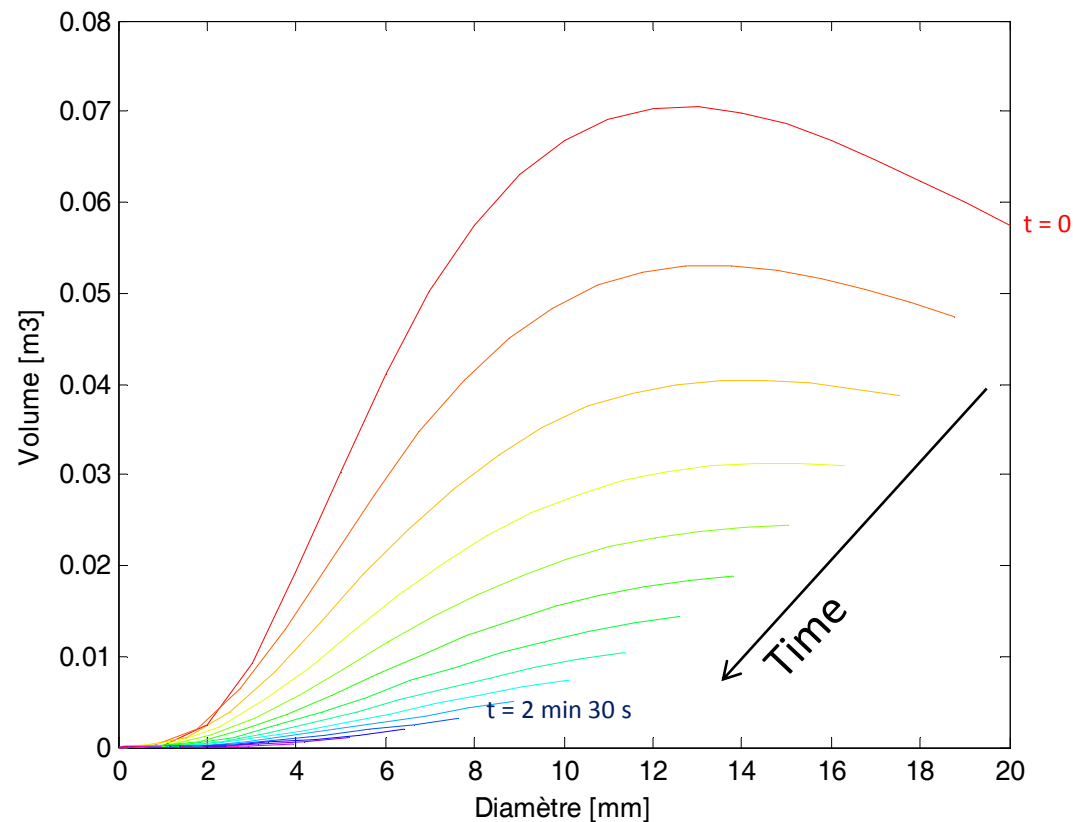
Example



Modelling a leak of butanol from a chemical tanker

1. Dissolution of the different droplet groups

- Evolution of volume over time ($\Delta t = 15$ s)



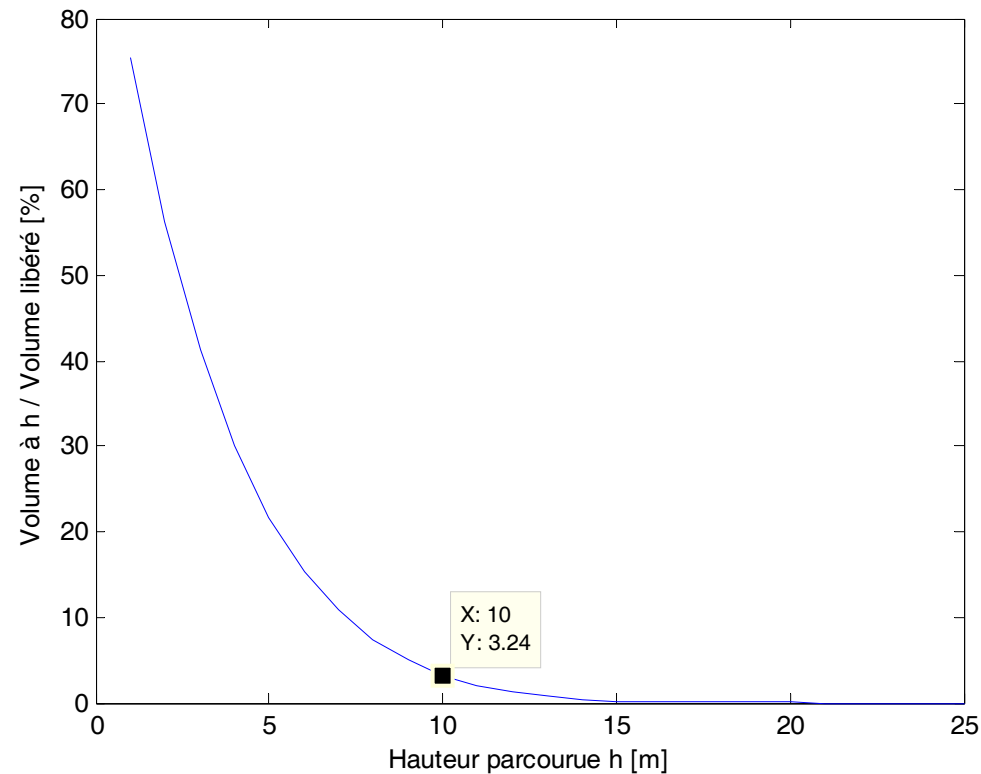


Example



Modelling a leak of butanol from a chemical tanker

Evolution of proportion of initial volume remaining with height



After 10 m: 96% of total volume is dissolved
After 20 m: the butanol is completely dissolved



The future



Modelling dispersion of chemicals at sea

Consideration of other mechanisms

Settling

Bioaccumulation

Volatilisation

Consideration of environmental parameters

Water temperature

Salinity

Hydrostatic pressure



**Thank you for
your attention**

